Grain Boundary Segregation and Mechanical Properties of an Aged Ni-20Cr-18W-1Mo Superalloy at Different Temperatures

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Abstract: The influence of aging temperature (200 to 800 °C) on grain boundary segregations in a Ni-20Cr-18W-1Mo superalloy was investigated by scanning electronic microscope (SEM), transmission electron microscopy (TEM), electron microprobe analyzer (EMPA) and a mechanical testing machine. Results indicate that the grain boundary segregation critical time of sulfur and phosphorus decreases with increasing of the aging temperature. Increasing of the aging temperature has a conspicuous effect on the concentration distribution in the grain boundary and the grain core. Grain boundary concentrations of sulfur and phosphorus increase with raising the testing temperature until a peak value is obtained at 650 and 400 °C, respectively, which is the essential reason to the declined mechanical properties from 200 to 600 °C.

Key words: Ni-20Cr-18W-1Mo superalloy; impurity elements; grain boundary segregation; mechanical property

Ni-Cr-W based superalloys are solid-solution strengthened alloys with more than 10 elements which are widely used in quest of excellent high-temperature strength, long-term creep rupture strength and corrosion resistance^[1-3]. They have found wide application in various heat resistant parts, such as combustor components of gas turbine engines, heating furnaces and components of chemical industries, especially heat exchanger tube of high temperature gas-cooled reactor^[4]. However, reliability and stability are the major issues of the nickel-base alloys used in steam generators^[5]. Additions of refractory elements have more effects on the microstructure and mechanical properties due to different elemental segregation behavior between a grain boundary and a grain interior. The changes in chemical composition, particularly sulfur and phosphorus, at grain boundaries inevitably affect intergranular failure, the fracture resistance and corrosion properties of many steels and superalloys, such as corrosion^[6,7] and stress corrosion cracking^[5,8,9]. Such grain boundary segregation may arise from heat treatment during fabrication or from diffusion processes during component service periods at elevated temperatures^[10].

The solid-solution strengthened Ni-20Cr-18W-1Mo superalloy contains Cr, W and Mo. During the melting process, impurity elements such as P and S are always doped in ingots. Sulfur has extremely low solid solubility ($<10^{-4}$) and a high-grain boundary enrichment ratio ($\sim 10^4$) to nickel-based alloys. Thus, sulfur significantly segregates at grain boundaries and induces embrittlement^[11]. Dong et al.^[12] have studied the stress-rupture life of Ni-base alloy 718 with different sulfur contents. The stress-rupture life dramatically decreased with the increasing sulfur content and obtained a minimum at 650 °C/686 MPa. Bruemmer et al.^[13] have found experimentally that sulfur is a more effective embrittling element of nickel alloys than phosphorus. Some researchers have indicated that phosphorus is a detrimental element accounting for the decrease of mechanical properties. Phosphorus promotes high-temperature cracking and hot tearing^[14]. It is well known that the effects of grain-boundary segregation on properties of alloys requires determining the segregation behavior of solute atoms experimentally^[15-17].

For the Ni-20Cr-18W-1Mo superalloy, extensive work has been focused on the carbide precipitates^[18-21], super-

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lattice precipitates^[22] and deformation characteristics^[23-25]. The researches on the effects of impurity elements segregation at the grain boundary have not been given any attention to. Therefore, the present work investigated the effects of impurities in Ni-20Cr-18W-1Mo superalloy on the segregation of the elements in the grain boundary. The evolution of elements segregation and their influence on the mechanical properties of the alloy were also analyzed.

1 Experiment

The chemical composition (wt%) of Ni-20Cr-18W-1Mo alloy used in this work was: Cr, 19.82; W, 18.48; Mo, 1.24; Al, 0.46; C, 0.11; B, 0.0028; La, 0.026; P, S <0.004, Bal. Ni. The process for fabrication of the wrought alloy was as follows. Firstly, the alloy was melted in a vacuum induction melting (VIM) furnace (with 1.3 MPa pressure) and remelted twice by vacuum arc remelting (VAR). Then, the ingot was homogenized at 1200 °C for 24 h and furnace cooled. Finally, the ingot was hot forged and rolled to a sheet with 9 mm thickness at 1150 °C. Then the sheet was annealed at 1260 °C for 0.5 h, followed by water quenching (WC). All the specimens for microstructure observation and mechanical testing were cut from the annealed sheet. Then they were aged at 200, 400, 600, 650, 700, 750 and 800 °C. The specimens were held for 20 min and ended with water quenching. SEM samples were prepared by standard metallographic techniques and the polished specimens were etched with aqua regia (HCl:HNO₃=3:1) for 120 s to reveal the microstructure. Elements concentrations were examined on EPMA-1720 tester equipped with SEM. Electron microprobe linear scans were used to measure the variations of each element.

For TEM, disks of 3 mm diameter were punched out from thin foils of 55 µm thickness. Electron transparent samples for TEM investigation were prepared by electrolytic jet polishing in a solution of 10 vol% perchloric acid in 90 vol% ethanol at -30 °C. TEM experiments were carried out using an FEI Tecnai G² F30 microscope operating at 300 kV.

Tensile tests were carried out using a Zink 150 mechanical testing machine with a strain rate maintained at $4 \times 10^{-4} \text{ s}^{-1}$ under air condition. The selected testing temperatures were 25, 200, 400, 600 and 650 °C. The samples were heated at the rate of 25 °C/min. When the selected temperature was reached, the samples were held for 20 min to make sure that a homogenous temperature distribution was achieved. After tensile tests, the samples were water quenched.

2 Results and Discussion

2.1 As-annealed microstructure

The microstructure of the as-annealed Ni-20Cr-18W-1Mo superalloy is shown in Fig.1, which is of typical fine

equiaxed grains. Grain boundaries and annealing twins are obvious in the microstructures (shown in Fig.1a). A small number of primary W-rich M₆C carbides randomly disperse inside the grains with various sizes. Fig.1b shows a typical prior austenite grain boundary. The selected image shows that there are no precipitates along the grain boundary. Moreover, the average size of grains is obtained by a point-intercept method according to the standard of ASTM E112-96. Eq.(1) shows the average grain size as follows:

$$\bar{l} = L / MP \quad (1)$$

where, $\bar{l}$ is average grain size, L is length of segment, M is the magnification of metallograph, and P is cut-off points number of every segment. The measured average grain size of the as-annealed Ni-20Cr-18W-1Mo superalloy is about 54.7 µm.

2.2 Critical time calculation

When a sample is maintained at a solution treatment temperature and then cooled to a certain lower temperature, lots of vacancies are absorbed by grain boundaries^[26]. The decrease in the vacancy concentration causes the decomposition of vacancy-solute complexes into vacancies and solute atoms. Therefore, a complex concentration gradient exists between the grain boundary and intracrystalline areas and leads the complexes to diffuse to the grain boundary. The diffusion causes supersaturated solute atoms to concentrate along the grain boundary. There must exist time when the complex diffusion flux to the grain boundaries balances the reverse solute diffusion flux. And the solute concentration of grain boundaries achieves a maximum value at this time. Therefore, the critical time is used to define the moment of maximum value in the isothermal process. After the solute concentration reaches peak, the de-segregation process is dominant. The level of solute concentration at grain boundaries decreases to the equilibrium concentration. The equation of critical time has been established by Xu^[27] based on the theory of Faulkner^[15] as follows:

$$t_c = \frac{r^2 \ln(D_c / D_i)}{\delta(D_c - D_i)} \quad (2)$$

where, D_i is diffusion coefficient for solute atom and D_c

Fig.1 SEM (a) and TEM (b) images of the as-annealed Ni-20Cr-18W-1Mo superalloy

is diffusion coefficient for vacancy-solute complexes, δ is the critical time constant and r is the grain radius.

Critical time is the moment that complex diffusion flux to the grain boundaries balances the reverse solute diffusion flux. The diffusion rates of complex and reverse solute are different with various temperatures. Therefore, the critical time changes with isothermal temperature. A calculation of critical time in the as-annealed Ni-20Cr-18W-1Mo superalloy can be obtained. The calculated parameters used in the critical time calculations are listed in Table 1.

Fig.2 shows the calculated relationship between critical time and aging temperatures for sulfur and phosphorus. It can be seen from Fig.2 that critical time of two elements decreases monotonically and rapidly with an increase in aging temperature. At the same aging temperature, critical time of phosphorus is much longer than that of sulfur. At 600 °C, phosphorus critical time is about 609 s, which is 10 times of that of sulfur. When the aging temperatures are lower than 600 °C, both the critical time of sulfur and phosphorus increases rapidly. They are 7095 and 785 s at 500 °C, 170865 and 20764 s at 400 °C, respectively, which are considerably longer than 1200s. When the aging temperatures are higher than 600 °C, the critical time of both elements decreases rapidly. They are 86 and 8.1 s at 700 °C, 4.7 and 0.4 s at 900 °C, which are considerably shorter than 1200 s.

2.3 Elements segregation at different aging temperatures

In order to study the elements segregation effects on the

Table 1 Data used in the theoretical calculations

Parameters	Phosphorus ^[6]	Sulfur ^[28]
$D_c/m^2 s^{-1}$	$4.23 \times 10^{-5} \exp(-1.4 \text{ eV}/kT)$	$4.23 \times 10^{-5} \exp(-1.4 \text{ eV}/kT)$
$D_i/m^2 s^{-1}$	$1.0 \times 10^{-8} \exp(-1.78 \text{ eV}/kT)$	$1.47 \times 10^{-4} \exp(-2.27 \text{ eV}/kT)$
δ	47	357
$r/\mu\text{m}$	27.35	27.35

Fig.2 Calculated relationship between critical time and aging temperatures for sulfur and phosphorus in as-annealed Ni-20Cr-18W-1Mo superalloy

properties of as-annealed Ni-20Cr-18W-1Mo superalloy, the linear scan analysis was accomplished by EMPA. Each linear scan contains two grains. Fig.3 shows the results of EMPA linear scan analysis of the aged samples with different temperatures. These curves have a minimum or a maximum value along the scan line for each element. Fig.3 gives the concentration profiles related to the crossing line of the grain boundary. The concentration profiles can give the width and variation of the enriched zone. According to EMPA linear scan of samples, the increasing of temperature has a conspicuous effect on the concentration distribution in the grain boundary and grain. The fluctuations of the linear scan analysis are increased for both elements by increasing the aging temperature up to 750 °C (Fig.3a~3g). But for the sample with 800 °C, the fluctuations of the linear scan decrease (Fig.3h). The grain boundary concentration of sulfur and phosphorus as a function of aging temperature is shown in Fig.4. Obviously, the grain boundary concentration of sulfur increases evidently with increasing the aging temperature until a peak value (0.0425 wt%) obtained at 650 °C, and then decreases drastically with further increasing aging temperature. The grain boundary concentration of phosphorus has a peak value (0.0220 wt%) at 400 °C.

In the model of Doig and Flewitt^[29,30], the kinetics is built by solving Fick's second law for the appropriate boundary conditions. The error function solution was used to analyze the solute concentration profile in the grain boundary region for a series of small time intervals of quenching.

$$(C_x - C_g)/(C_b - C_g) = \text{erfc}\left\{x/\left[2(D_c t)^{1/2}\right]\right\} \quad (3)$$

where, C_x is the solute concentration at a distance x (0~3 μm) from the grain boundary, C_b is the solute concentration at the grain core at the quenching temperature for the quenching time t (30 s), C_g is the boundary vacancy concentration after zero time and D_c is the diffusion coefficient for the complexes.

Fig.5 shows the calculated kinetic curves of sulfur and phosphorus from Eq.(3) with variation of distances from grain boundary. The dash lines stand for the position of grain boundary, which show that sulfur and phosphorus are both segregated to the grain boundaries at 400 °C in a layer about 2 μm . As mentioned above, the concentration of solute depends on how the critical time approaches the aging time. When the aging temperature changes from 400 °C to 600 °C, the segregated degree of sulfur increases while that of phosphorus decreases dramatically. The aging temperature appears to have the greatest impact on the segregation of sulfur and phosphorus. It can be seen from Fig.5 that the kinetic curve of sulfur at 600 °C is more slowdown than that at 400 °C. On the contrary, the grain boundary concentration of phosphorus at 400 °C is higher

Fig.3 Depth compositional profiles of S and P near the grain boundary aged at different temperatures for 1200 s: (a) 20 °C, (b) 200 °C, (c) 400 °C, (d) 600 °C, (e) 650 °C, (f) 700 °C, (g) 750 °C, and (h) 800 °C

Fig.4 Grain boundary concentration of sulfur and phosphorus as a function of aging temperature

Fig.5 Calculated kinetic curves of sulfur and phosphorus from Eq.(3) with variation of distances from grain boundary

than that at 600 °C. The results of EMPA and kinetic calculation indicate that sulfur and phosphorus are the main grain boundary segregation elements in the experimental Ni-20Cr-18W-1Mo superalloy from 25 °C to 600 °C.

2.4 Mechanical properties

Mechanical properties of superalloys have to be considered during the hot working and service process, which are always related to chemical compositions and precipitations on grain boundaries. Tensile tests are used in this work to evaluate the mechanical properties of experimental alloys. Tensile property variation of Ni-20Cr-18W-1Mo superalloy as a function of aging temperature is illustrated in Fig.6. The results show that the tensile strength decreases from 855 to 710 MPa with the increase of aging temperature. Previous research^[31] has proved that there were differences in the volume fraction of

carbides after heat-treatments above 650 °C and the carbides can influence the mechanical properties of superalloys. The mechanical properties of the samples are affected by carbide's size, volume fraction and interparticle spacing^[32]. Combined with elongation results which are displayed in Fig.6 along with tensile strength, the elongation decreases with the increasing test temperature from 25 °C to 600 °C. When the temperature is higher than 600 °C, the elongation increases rapidly. It shows a maximum embrittlement at 600 °C and the ductility minimum is about 43% (RA%). Fig.7 shows the fracture morphologies of tensile fractured specimens. In the low temperature condition, the fracture mode is the mixture of the trans- and inter-granular (Fig.7a). However, as the concentrations of sulfur are extensively riched in the grain boundary region at 400 °C, the inter-granular fracture mode

Fig.6 Effect of temperature on tensile properties of as-annealed Ni-20Cr-18W-1Mo superalloy

is dominant in the tensile samples (Fig.7b). For the sample tested at 600 °C, some large precipitates-induced cracks are emerge as shown in Fig.7c, which is consistent with the previous tensile test results by Bai et al.^[31]. By comparing Fig.7a, 7b and 7c, it can be found that the density of dimples on the fracture surfaces for the samples aged at 600 °C is larger than that for the samples aged at 200 and 400 °C. During the hot deformation, the existence of carbide can cause the stress concentration around the carbide particles. As the stress reaches the bond strength of the matrix and carbide, microvoids nucleate^[33]. From the fracture microstructures, it is obvious that the concentrations of sulfur and phosphorus affect mechanical properties depending on aging temperature. It is seen from Fig.3 and Fig.4 that the maximum of the grain boundary concentration for sulfur and phosphorus corresponds to the low ductility

Fig.7 Fracture morphologies of tensile tested Ni-20Cr-18W-1Mo superalloy at different temperatures: (a) 200 °C, (b) 400 °C, and (c) 600 °C

for the alloy at 400 °C. When the test temperatures are lower than 400 °C, the grain-boundary concentration of sulfur and phosphorus and the degree of embrittlement of this alloy increase with the increasing the test temperature. When the test temperatures range from 400 °C to 600 °C, the grain-boundary concentration of sulfur increases while that of phosphorus decreases, and the degree of embrittlement also increases with increasing the test temperature. The increase in bulk sulfur contents enhances the sensitivity to intergranular embrittlement^[26]. Combined with the critical time theory and mechanical tensile test, the concentration of sulfur and phosphorus and the degree of embrittlement depend on how the critical time approaches the aging time. Fig.2 shows that the critical time of sulfur at 400 °C to 600 °C is closer to the aging time of 1200 s than that at other aging temperatures. Therefore, both the grain-boundary concentration of 600 °C and the degree of embrittlement can attain maxima for aging for 1200 s at 600 °C.

Fig.8 shows the SEM microscopy of samples after aging testing at different temperatures. It can be clearly seen from Fig.8a and 8b that no dynamic recrystallization occurs

during the tensile test. It is also clear that no precipitates exist at grain boundaries. XRD analysis (Fig.9) identifies that the phase compositions are chromium-rich γ phase and tungsten-rich M_6C -type carbides. Bai et al.^[31] have indicated that the lamellar $M_{23}C_6$ -type carbides which precipitate at 600 °C induce premature failure of Ni-20Cr-18W-1Mo superalloy. At the temperatures ranging from 600 °C to 900 °C, the anomalous yield strength could be attributed to the formation of discrete $M_{23}C_6$ -type carbides. From Fig.8c, 8d, 8e and 8f, lots of second phase precipitates can be found along the grain boundaries after tests at temperatures ranging from 600 °C to 800 °C. Maximum embrittlement in the elevated temperature tension tests which is induced by a maximum in the grain-boundary concentration of sulfur at 500 °C has been found in experimental Ni-Cr-Fe alloy^[28]. Sulfur grain boundary segregation during hot deformation at 450 and 500 °C of nickel (before recrystallization) has been also indicated by Allart et al.^[34]. At low temperatures (< 600 °C), the segregation of alloying elements (such as W, Cr, Mo) does not have enough energy. It can be indicated that low ductility which is induced by sulfur and phosphorus

Fig.8 SEM microstructures of as-aged Ni-20Cr-18W-1Mo superalloy at different temperatures: (a) 200 °C, (b) 400 °C, (c) 600 °C, (d) 650 °C, (e) 700 °C and (f) 800 °C

Fig.9 XRD patterns of the as-annealed Ni-20Cr-18W-1Mo superalloy aged at different temperatures for 1200 s

grain boundary segregation is widespread in Ni-based alloys. If more than one kind of impurity segregates to the grain boundaries, site competition and chemical interaction among impurities at grain boundaries occur^[35]. During isothermal holding at 1260 °C for 1800 s, both sulfur and phosphorus completely segregate to the grain boundaries, and the concentrations at grain boundaries are various and approach equilibrium values. When the sample is held at 600 °C for 1200 s before the tensile test, the grain boundary concentration of the impurities shows different levels. Combined with Fig.5, the kinetic curves show that the concentration of sulfur and phosphorus increase with the increasing the testing temperatures. It also indicates that the grain boundary concentration of sulfur is more than that of phosphorus at 600 °C. From the above, the ductility of

Ni-20Cr-18W-1Mo superalloy decreases with the increasing test temperature from 25 °C to 600 °C, which is ascribed to grain boundary segregation of sulfur and phosphorus. The impurities segregation and ductility variations could be reasonably explained by impurity elements grain boundary segregation, site competition and chemical interaction.

3 Conclusions

1) The critical time of each element in Ni-20Cr-18W-1Mo superalloy decreases monotonically and rapidly with the increase in aging temperature. However, critical time of phosphorus is much longer than that of sulfur at the same aging temperature.

2) The grain boundary concentration of sulfur increases with the increasing aging temperature until a peak value (0.0425 wt%) at 650 °C, and then decreases with the further increasing aging temperature. The grain boundary concentration of phosphorus has a peak value (0.022 wt%) at 400 °C.

3) The tensile strength and elongation values of samples decrease as the temperature increases from 25 °C to 600 °C. The tensile strength decreases from 855 MPa to 710 MPa and the ductility minimum is about 43% (RA%). In addition, the increased grain boundary concentration of sulfur and phosphorus accounts for the reduced ductility of Ni-20Cr-18W-1Mo superalloy from 200 °C to 600 °C.

References

- 1 Pataky G J, Sehitoglu H, Maier H J. *Materials Characterization*[J], 2013, 75: 69
- 2 Chen X, Yang Z Q, Sokolov M A et al. *Materials Science and Engineering A*[J], 2013, 563: 152

- 3 Tung H M, Stubbins J F. *Materials Science and Engineering A*[J], 2012, 538: 1
- 4 Liu Y, Hu R, Li J S et al. *Materials Science and Engineering A*[J], 2008, 497(1-2): 283
- 5 Zinkle S J, Was G S. *Acta Materialia*[J], 2013, 61(3): 735
- 6 Wang K, Si H, Yang C et al. *Journal of Iron and Steel Research International*[J], 2011, 18(1): 61
- 7 Xia S, Li H, Liu T G et al. *Journal of Nuclear Materials*[J], 2011, 416(3): 303
- 8 Briant C L. *Metallurgical and Materials Transactions A*[J], 1988, 19(1): 137
- 9 Caceras P, Ralph B, Allen G C et al. *Surface and Interface Analysis*[J], 1988, 12(3): 191
- 10 Ding R. *Materials Science and Technology*[J], 2010, 26(1): 36
- 11 Kobayashi S, Tsurekawa S, Watanabe T et al. *Scripta Materialia*[J], 2010, 62(5): 294
- 12 Dong J X, Xie X S, Thompson R G. *Metallurgical and Materials Transactions A*[J], 2000, 31(9): 2135
- 13 Bruemmer S M, Jones R H, Thomas M T et al. *Metallurgical Transactions A*[J], 1983, 14(1): 223
- 14 Song X L, Yuan Z X, Jia J et al. *Journal of Materials Science and Technology*[J], 2010, 26(9): 793
- 15 Faulkner R G. *Journal of Materials Science*[J], 1981, 16(2): 373
- 16 Song S H, Xu T D, Yuan Z X. *Acta Metallurgica*[J], 1989, 37(1): 319
- 17 Xu T D, Song S H. *Acta Metallurgica*[J], 1989, 37(9): 2499
- 18 Hu R, Bai G H, Li J S et al. *Materials Science and Engineering A*[J], 2012, 548: 83
- 19 Bai G H, Li J S, Hu R et al. *Materials Science and Engineering A*[J], 2011, 528(6): 2339
- 20 Bai G H, Hu R, Li J S et al. *Rare Metal Materials and Engineering*[J], 2011, 40(10): 1737 (in Chinese)
- 21 Han Y B, Xue X Y, Zhang T B et al. *Materials Science and Engineering A*[J], 2016, 667: 391
- 22 Gao X Y, Hu R, Zhang T B et al. *Materials Characterization*[J], 2016, 111: 86
- 23 Liu Y, Hu R, Li J S et al. *Materials Science and Engineering A*[J], 2009, 508(1-2): 141
- 24 Bai G H, Li J S, Hu R et al. *Rare Metal Materials and Engineering*[J], 2011, 40(7): 1300 (in Chinese)
- 25 Han Y B, Xue X Y, Zhang T B et al. *Rare Metal Materials and Engineering*[J], 2015, 44(9): 2314 (in Chinese)
- 26 Xu T D, Cheng B Y. *Progress in Materials Science*[J], 2004, 49(2): 109
- 27 Xu T D. *Journal of Materials Science*[J], 1987, 22(1): 337
- 28 Wang K, Xu T D, Wang Y Q et al. *Philosophical Magazine Letters*[J], 2009, 89(11): 725
- 29 Doig P, Flewitt P E J. *Metallurgical Transactions A*[J], 1987, 18(3): 399
- 30 Doig P, Flewitt P E J. *Acta Metallurgica*[J], 1981, 29(11): 1831
- 31 Bai G H, Li J S, Hu R et al. *Materials Science and Engineering A*[J], 2011, 528(4-5): 1974
- 32 Rahimian M, Milenkovic S, Sabirov I. *Journal of Alloys and Compounds*[J], 2013, 550(4-5): 339
- 33 Xu Z F, Jiang L, Dong J S et al. *Journal of Alloys and Compounds*[J], 2015, 620: 197
- 34 Allart M, Christien F, Le Gall R. *Acta Materialia*[J], 2013, 61(20): 7938
- 35 Zhang Z L, Lin Q Y, Yu Z S. *Materials Science and Technology*[J], 2000, 16(3): 305

不同温度时效 Ni-20Cr-18W-1Mo 合金晶界偏聚及力学性能研究

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摘 要: 采用扫描电镜 (SEM)、透射电镜 (TEM)、电子探针 (EPMA) 和高温力学试验机等手段, 研究了不同时效温度 (200~800 °C) 对 Ni-20Cr-18W-1Mo 高温合金的元素晶界偏聚和力学性能的影响。结果表明, 硫、磷元素的晶界偏聚临界时间随时效温度升高而缩短; 时效温度对元素在晶界和晶内的成分分布有显著的影响; 实验合金的抗拉强度和延伸率随时效温度升高而降低。分析发现, 硫、磷元素在晶界中的含量随时效温度升高而增大直至两者分别在 650 和 400 °C 时达到峰值, 是合金在 200~600 °C 区间力学性能降低的重要原因。

关键词: Ni-20Cr-18W-1Mo 高温合金; 杂质元素; 晶界偏聚; 力学性能

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